

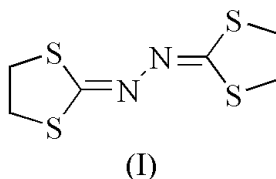
1,2-Bis(1,3-dithiolan-2-ylidene)hydrazine

Liang-Zhong Xu,^{a*} Hai-Zhen Xu,^b
Shuang-Hua Yang,^a Chun-Li Li^a
and Kai Zhou^a^aCollege of Chemistry and Molecular Engineering, Qingdao University of Science and Technology, Qingdao 266042, People's Republic of China, and ^bCollege of Chemistry and Life Science, Tianjin, Normal University, Weijin Road No. 241, Tianjin, People's Republic of ChinaCorrespondence e-mail:
youquan_zhu@mail.nankai.edu.cn

Key indicators

Single-crystal X-ray study
 $T = 293$ K
Mean $\sigma(\text{C}-\text{C}) = 0.005$ Å
 R factor = 0.046
 wR factor = 0.130
Data-to-parameter ratio = 17.7For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.The title molecule, $\text{C}_6\text{H}_8\text{N}_2\text{S}_4$, possesses a crystallographically imposed center of symmetry. The two five-membered rings are in half-chair conformations. The crystal packing is stabilized mainly by van der Waals forces.Received 19 November 2004
Accepted 30 November 2004
Online 4 December 2004

Comment

As an important type of fungicide, heterocyclic compounds are highly efficient in view of their low toxicity (Shi *et al.*, 1995; Xu *et al.*, 2002). In this paper, we report the crystal structure of the title compound, (I) (Fig. 1).In the solid state, the molecule of (I) possesses a crystallographically imposed center of symmetry. All bond lengths and angles are normal (Table 1). The 1,3-dithiolane rings are each in a half-chair conformation. The deviations of atoms C2 and C3 from the plane formed by S1, S2 and C1 are 0.226 (7) and -0.442 (6) Å, respectively. The crystal packing (Fig. 2) is stabilized mainly by van der Waals forces.

Experimental

The title compound, (I), was prepared according to the method of Mayer & Schaefer (1964) and was crystallized from a mixture of ethanol and petroleum ether.

Crystal data

$\text{C}_6\text{H}_8\text{N}_2\text{S}_4$	$D_x = 1.619$ Mg m ⁻³
$M_r = 236.38$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/n$	Cell parameters from 1494 reflections
$a = 6.300$ (2) Å	$\theta = 2.6\text{--}26.4^\circ$
$b = 7.737$ (2) Å	$\mu = 0.92$ mm ⁻¹
$c = 10.386$ (3) Å	$T = 293$ (2) K
$\beta = 106.712$ (4) $^\circ$	Prism, colorless
$V = 484.9$ (2) Å ³	$0.30 \times 0.28 \times 0.22$ mm
$Z = 2$	

Data collection

Bruker SMART CCD area-detector diffractometer	992 independent reflections
φ and ω scans	834 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan, (SADABS; Sheldrick, 1996)	$R_{\text{int}} = 0.023$
$T_{\text{min}} = 0.708$, $T_{\text{max}} = 0.816$	$\theta_{\text{max}} = 26.4^\circ$
2712 measured reflections	$h = -7 \rightarrow 7$
	$k = -9 \rightarrow 9$
	$l = -12 \rightarrow 8$

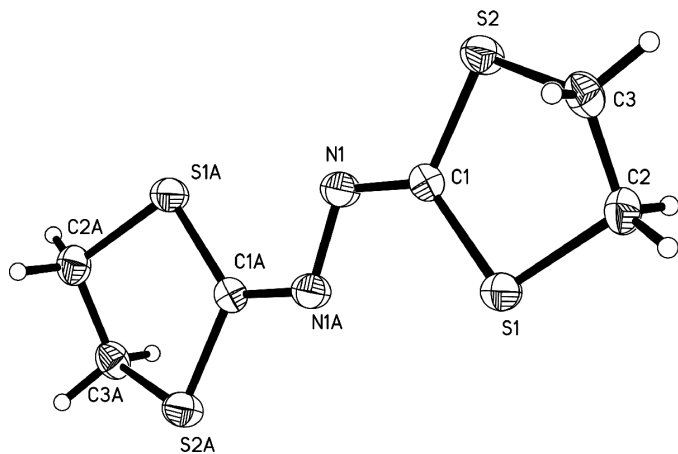


Figure 1
View of the molecule of (I), with displacement ellipsoids drawn at the 30% probability level. The suffix A corresponds to symmetry code (i) in Table 1.

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.046$
 $wR(F^2) = 0.130$
 $S = 1.04$
 992 reflections
 56 parameters
 H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0675P)^2 + 0.6306P]$$

where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.0001$
 $\Delta\rho_{\max} = 1.06 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.37 \text{ e } \text{\AA}^{-3}$
 Extinction correction: *SHELXTL*
 Extinction coefficient: 0.161 (16)

Table 1

Selected geometric parameters (\AA , $^\circ$).

S1—C1	1.757 (3)	N1—C1	1.287 (4)
S1—C2	1.816 (3)	N1—N1 ⁱ	1.396 (5)
S2—C1	1.749 (3)	C2—C3	1.506 (5)
S2—C3	1.803 (3)		
C1—S1—C2	95.5 (2)	N1—C1—S1	125.8 (2)
C1—S2—C3	94.9 (2)	S2—C1—S1	115.3 (2)
C1—N1—N1 ⁱ	111.2 (3)	C3—C2—S1	108.1 (2)
N1—C1—S2	118.9 (2)	C2—C3—S2	108.8 (2)
N1 ⁱ —N1—C1—S2	179.1 (3)	C2—S1—C1—S2	-7.2 (2)
N1 ⁱ —N1—C1—S1	-2.6 (5)	C1—S1—C2—C3	31.4 (3)
C3—S2—C1—N1	164.3 (3)	S1—C2—C3—S2	-45.2 (3)
C3—S2—C1—S1	-14.2 (2)	C1—S2—C3—C2	36.0 (3)
C2—S1—C1—N1	174.4 (3)		

Symmetry codes: (i) $-x + 2, -y, -z$.

All H atoms were placed in calculated positions, with C—H distances of 0.97 \AA , and included in the final cycles of refinement

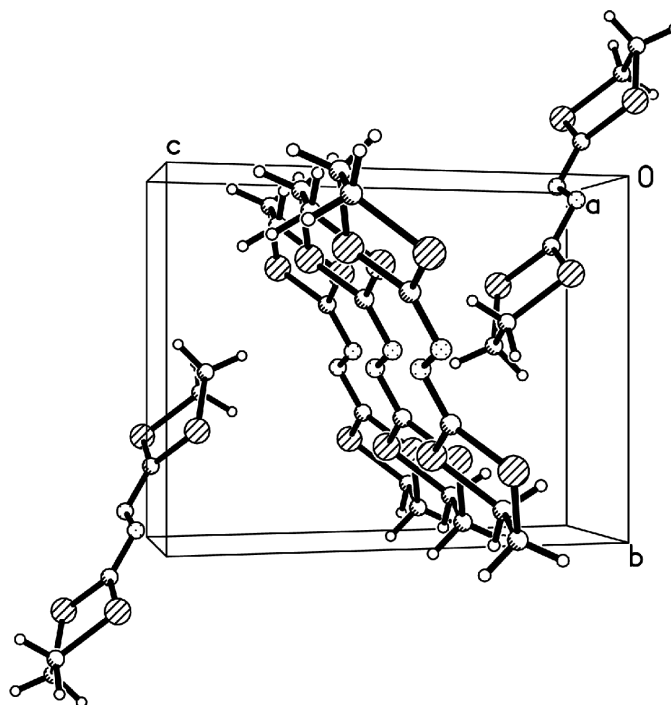


Figure 2
Packing diagram of the title compound, viewed along the *a* axis.

using a riding model, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$. The maximum electron-density peak was located 0.83 \AA from atom S2.

Data collection: *SMART* (Bruker, 1998); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1999); software used to prepare material for publication: *SHELXTL*.

References

- Bruker (1998). *SMART*. Bruker AXS Inc., Madison, Wisconsin, USA.
 Bruker (1999). *SAINT* and *SHELXTL*. Bruker AXS Inc., Madison, Wisconsin, USA.
 Mayer, R. & Schaefer, K. (1964). *J. Prakt. Chem.* **26**, 279–295.
 Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
 Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
 Shi, Y. N., Lu, Y. C. & Fang, J. X. (1995). *Chem. J. Chin. Univ.* **16**, 1710–1713.
 Xu, L. Z., Zhang, S. S., Li, H. J. & Jiao, K. (2002). *Chem. Res. Chin. Univ.* **18**, 284–286.